

## ABIETANE DITERPENOIDS FROM THE ROOTS OF *SALVIA CRYPTANTHA*

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**Key Work Index**—*Salvia cryptantha*; Labiatae; diterpenoids; horminone; 7-acetylhorminone; 2 $\beta$ -hydroxyroyleanone; cryptanol (11,12,14-trihydroxyabietae-6,8,11,13-tetraen).

**Abstract**—In addition to the known compounds horminone and 7-acetylhorminone, two new abietane diterpenoids, 2 $\beta$ -hydroxyroyleanone and cryptanol, were isolated from the roots of *Salvia cryptantha*.

### INTRODUCTION

The aerial parts of *Salvia* species contain several triterpenoid compounds [1–6]. However, only two species, *S. triloba* [7] and *S. tomentosa* [8], have given diterpenoids. As a continuation of our investigations, we have examined the roots of *Salvia cryptantha* Montbret and Auches ex. Benth for diterpenoids. The possible pharmacological activities of *S. cryptantha* will be the subject of another study.

### RESULTS AND DISCUSSION

We isolated four diterpenoids from the roots of *S. cryptantha*, two of which are new compounds. The known diterpenoids horminone (7 $\beta$ -hydroxyroyleanone) (1) and 7-acetylhorminone (7 $\beta$ -acetoxyroyleanone) (2) were previously isolated from *S. lanata* [9]. The structures of the known compounds were established by comparing their spectral data to literature values.

Cryptanol (3) has the composition  $C_{20}H_{28}O_3$  on the basis of mass spectrometry ( $[M]^+ m/z$  316, 44%). Its IR spectrum showed bands at 3350  $cm^{-1}$  for a hydroxyl group, aromatic bands at 1600, 1550, 1535  $cm^{-1}$  and conjugation at 1635  $cm^{-1}$ ; the absence of the bands at 1665 and 1645  $cm^{-1}$  indicated an aromatic rather than an *ortho*- or *para*-quinone ring system. Its UV spectrum ( $\lambda_{max}$  332 and 274 nm) indicated conjugation of an aromatic ring to a double bond rather than conjugation between a *para*-quinoid system and a double bond as in dehydoroyleanone [10]. Its  $^1H$  NMR spectrum showed three methyl signals as singlets at  $\delta$ 0.87 (C-4 $\alpha$ ), 0.91 (C-4 $\beta$ ) and 1.27 (C-10), and two methyl doublets for an isopropyl group at  $\delta$ 1.0 and 1.05 ( $J = 7$  Hz for each,  $-HC \begin{smallmatrix} \diagup CH_3 \\ \diagdown CH_3 \end{smallmatrix}$ ); the isopropyl methine proton was at  $\delta$ 3.14 (septet,  $J = 7$  Hz). All these signals are common for methyl groups in the abietan skeleton [10–12]. The other protons were as follows: a double-doublet at  $\delta$ 6.45 ( $J = 3$  and 9 Hz, H-6) and 6.79 ( $J = 3$  and 9 Hz, H-7) forming the AB part of an ABX system indicated a double bond in the molecule, the X part being at  $\delta$ 2.13 (t,  $J = 3$  Hz, H-5). There are three possible positions for the double bond: C-1, C-2 or C-6.

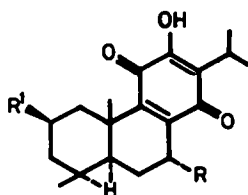
Since there is conjugation between the double bond and the aromatic ring system as shown by its UV spectrum (332 nm), the only possible position for the double bond is at C-6. This was confirmed by COSY experiments. The absence of any signals between  $\delta$ 3.14 and 6.45 indicated that the hydroxyl groups were in the aromatic ring rather than in rings A and B. The signals at  $\delta$ 7.3, 7.24 (under the  $CHCl_3$  peak) and at  $\delta$ 7.2 indicated three phenolic hydroxyl groups. All these data showed the structure of cryptanol as given in 3.

The mass peak of 2 $\beta$ -hydroxyroyleanone (4) indicated the molecular formula  $C_{20}H_{28}O_4$  ( $[M]^+ m/z$  332, 14%). Its IR spectrum showed a hydroxyl band at 3420  $cm^{-1}$  and *para*-quinone bands at 1665, 1640 and 1610  $cm^{-1}$ . In addition, the UV spectrum ( $\lambda_{max}$  399, 272 nm) indicated the presence of a *para*-quinone group. The  $^1H$  NMR spectrum of 4 showed the methyl signals as singlets at  $\delta$ 0.88 (C-4 $\alpha$ ), 0.9 (C-4 $\beta$ ) and 1.27 (C-10), and two methyl doublets at  $\delta$ 1.25 ( $J = 7$  Hz, for each  $-HC \begin{smallmatrix} \diagup CH_3 \\ \diagdown CH_3 \end{smallmatrix}$ ); the isopropyl methine proton was at  $\delta$ 3.14 (septet,  $J = 7$  Hz). A  $H_1$  triple doublet at  $\delta$ 4.2 ( $J = 1.1$ , 2.5 and 4.5 Hz) indicated the presence of a secondary alcohol group having two methylene groups in the structure; the coupling constants established the equatorial position of the proton at C-2. After comparing the chemical shifts and splitting pattern to those of 6 $\alpha$ , 6 $\beta$ , 7 $\alpha$ , 7 $\beta$ , 2 $\alpha$  and 3 $\beta$ -hydroxyroyleanones [10, 12, 13] the structure of 4 is established as 2 $\beta$ -hydroxyroyleanone.

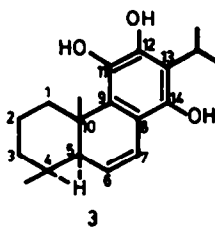
### EXPERIMENTAL

Roots of *Salvia cryptantha*, collected from Afyon (western Turkey) in July 1982, were identified by Dr. E. Tuzlaci (University of Marmara, Istanbul). A voucher specimen (ISTE 48915) has been deposited at the Herbarium of the Faculty of Pharmacy, University of Istanbul.

**Extraction and isolation of the compounds.** Air-dried and powdered roots (370 g) were extracted with petrol,  $Me_2CO$  and EtOH, respectively at room temp. The solvents were removed in



- 1 R=OH  
 R'=H  
2 R=OAc  
 R'=H  
4 R=H  
 R'=OH



*vacuo*. The petrol concentrate (2 g) when separated on a silica gel column (2 x 50 cm) yielded diterpenoids, which were further purified and/or separated by prep. TLC. Elution was started with petrol; a gradient of EtOAc was added up to 100%. The 5–20% EtOAc eluate yielded, in order of elution, cryptanol (20 mg), horminone (15 mg), 7-acetylhorminone (12 mg) and 2 $\beta$ -hydroxyroyleanone (10 mg). Me<sub>2</sub>CO and EtOH extracts contained almost the same compounds together with triterpenoids and flavonoids.

**Cryptanol (3).** Mp 138–142°; IR  $\nu_{\text{max}}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 3350, 2970, 2920, 2840, 1635, 1600, 1550, 1535, 1455, 1435, 1405, 1385, 1375, 1330, 1270, 1250, 1155, 1100, 960, 900, 800, 770, 755, 735, 720; UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 332 (log  $\epsilon$  2.9), 274 (log  $\epsilon$  4.2); <sup>1</sup>H NMR (FT-NT,

300 MHz, CDCl<sub>3</sub>); see text; MS (probe, 70 eV, Varian MAT 311)  $m/z$  (rel. int.): 316 [M]<sup>+</sup> (44), 301 [M – 15]<sup>+</sup> (7.5), 279 (8.8), 271 (2.3), 258 (2.7), 245 (9.7), 232 (11.6), 231 (15.5).

**2 $\beta$ -Hydroxyroyleanone (4).** IR  $\nu_{\text{max}}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 3420, 2970, 2925, 2860, 1665, 1640, 1610, 1460, 1430, 1395, 1375, 1330, 1250, 1170, 1150, 1130, 1060, 1025, 960; UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 399 (log  $\epsilon$  2.8), 272 (log  $\epsilon$  4.2); <sup>1</sup>H NMR (FT-NT, 300 MHz, CDCl<sub>3</sub>) given in the text; MS (probe, 70 eV, Varian MAT 311)  $m/z$  (rel. int.): 332 [M]<sup>+</sup> (14), 314 [M – 18]<sup>+</sup> (18.1), 299 [M – 18 – 15]<sup>+</sup> (3.5), 245 (7.4), 231 (9.6).

#### REFERENCES

1. Ulubelen, A. and Ayanöglu, E. (1975) *Phytochemistry* **14**, 309.
2. Ulubelen, A. and Brieskorn, C. H. (1975) *Phytochemistry* **14**, 820.
3. Ulubelen, A., Brieskorn, C. H. and Özdemir, N. (1977) *Phytochemistry* **16**, 790.
4. Ulubelen, A., Miski, M. and Mabry, T. J. (1981) *J. Nat. Prod.* **44**, 586.
5. Ulubelen, A. and Topçu, G. (1984) *J. Nat. Prod.* **47**, 1068.
6. Ulubelen, A., Miski, M., Johansson, C., Lee, E., Mabry, T. J. and Matlin, S. (1985) *Phytochemistry* **24**, 1386.
7. Ulubelen, A., Öztürk, S. and Isildatici, S. (1968) *J. Pharm. Sci.* **57**, 103.
8. Ulubelen, A., Miski, M. and Mabry, T. J. (1981) *J. Nat. Prod.* **44**, 119.
9. Mukherjee, K. S., Ghosh, P. K. and Badruddoza, S. (1981) *Phytochemistry* **20**, 1441.
10. Hensh, M., Rüedi, P. and Eugster, C. H. (1975) *Helv. Chim. Acta* **58**, 1921.
11. Miyase, T., Rüedi, P. and Eugster, C. H. (1977) *Helv. Chim. Acta* **60**, 272.
12. Rüedi, P. (1984) *Helv. Chim. Acta* **67**, 1116.
13. Hayashi, T., Handa, T., Ohashi, M. and Kakisawa, H. (1971) *Chem. Commun.* **304**, 541.